

Solid, liquid and gas

solids	strong	liquids	gases	compressed	weak
(a) one shape. There	are hard and can			heavy. They cannot f place.	low so stay
(d) holding them but	can flow. Their shat they are not as (f)			(e)The	ere are bor
	TO	pace. They are light between the partion		about. They can be .	(i)
ls each of these s	statements true or	false? Circle your	choice.		
(a) Compounds	always contain dif	ferent elements.			True
(b) The particles in a solid can take the shape of any container.					True
(c) Particles in a liquid can move around.					True
(d) Particles in a gas have the most kinetic energy.					True
(a) Most motals	er er er er er er				
(e) Most metals	are solids.				True
	are solids. ow, draw the arran	gement of particle	s in a solid, liquid	d and gas.	True
		gement of particle	s in a solid, liquid	d and gas.	True,
In the boxes belo	ow, draw the arran	1	s in a solid, liquid	175. 175. 175. 175. 175. 175. 175. 175. 175.	True
In the boxes belo		1	s in a solid, liquid	175. 175. 175. 175. 175. 175. 175. 175. 175.	True,
In the boxes belo	ow, draw the arran	1	s in a solid, liquid	175. 175. 175. 175. 175. 175. 175. 175. 175.	True,
In the boxes belo	ow, draw the arran	1	s in a solid, liquid	175. 175. 175. 175. 175. 175. 175. 175. 175.	True
In the boxes belo	ow, draw the arran	(b) Liquid	•	(c) Gas	
In the boxes belo	ow, draw the arran	(b) Liquid	ave a fixed shape	175. 175. 175. 175. 175. 175. 175. 175. 175.	
In the boxes belo	ow, draw the arrange	(b) Liquid	ave a fixed shape	(c) Gas	
In the boxes belo	ow, draw the arrange	(b) Liquid	ave a fixed shape	(c) Gas	
In the boxes belo (a) Solid It is observed the this substance contact the co	at a particular substan be either a liqui	(b) Liquid stance does not had or a gas but not	ave a fixed shape a solid.	(c) Gas	

In your book, explain why, when you have a shower, water can appear on the bathroom mirror.

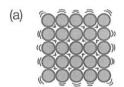


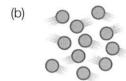
Solid, liquid and gas

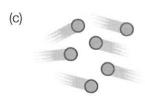
1.	In the boxes below,	draw the arrangement	of particles in	n a solid,	liquid and gas.
----	---------------------	----------------------	-----------------	------------	-----------------

(a) Solid	(b) Liquid	(c) Gas
		•

2.	Write captions for t	the water molecules below to	summarise what is	happening at the	particle level.
----	----------------------	------------------------------	-------------------	------------------	-----------------







- 3. Answer these questions in your book.
 - (a) Explain why, when you have a shower, water can appear on the bathroom mirror.
 - (b) Explain why clothes will dry faster on a windy day compared with clothes on a day without wind.
- 4. Often when you get a can of soft drink from the fridge, beads of water (condensation) appear on the outside of the can soon after. Explain why this is.
- 5. When you exercise, your body produces perspiration (sweat) to cool you down. Explain why your body does this.